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PPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/847,345	05/03/2001	Abraham Benderly	A01040A	6887
21898	7590 05/06/2003			
ROHM AND HAAS COMPANY PATENT DEPARTMENT 100 INDEPENDENCE MALL WEST PHILADELPHIA, PA 19106-2399		EXAMINER		
			LANGEL, V	VAYNE A
			ART UNIT	PAPER NUMBER
			1754	P
			DATE MAILED: 05/06/2003	6

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No. 8473	45 1	Applicant(s)	enderly	etal
· Office Action Summary	Examiner	ang.	e/	Group Art Unit	
-The MAILING DATE of this communication app	cears on the cover sh	neet bene	ath the co	orrespondence ad	dress—
Period for Reply	,				
A SHORTENED STATUTORY PERIOD FOR REPLY IS SI OF THIS COMMUNICATION.	ET TO EXPIRE	<u>ح</u>	MONTH(S	FROM THE MAII	LING DATE
 Extensions of time may be available under the provisions of 37 from the mailing date of this communication. If the period for reply specified above is less than thirty (30) day If NO period for reply is specified above, such period shall, by a Failure to reply within the set or extended period for reply will, It Any reply received by the Office later than three months after the term adjustment. See 37 CFR 1.704(b). 	ys, a reply within the statut default, expire SIX (6) MON by statute, cause the appli	tory minimu VTHS from t ication to be	ım of thirty (3 the mailing d ecome ABAF	30) days will be consid late of this communica NDONED (35 U.S.C. §	ered timely. tion. 133).
Status					
☐ Responsive to communication(s) filed on					·
☐ This action is FIMAL.					
 Since this application is in condition for allowance ex accordance with the practice under Ex parte Quayle, 			cution as (to the merits is cl	osed in
Disposition of Claims					
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U.S. Patent and Trademark Office PTO-326 (Rev. 11/00)

Part of Paper No.

Serial No. 09/847,345
Art Unit 1754

The following is a quotation of the appropriate paragraphs of 35 U.S.C. § 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless --

- (b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.
- (e) The invention was described in (1) an application for patent, published under Section 122(b), by another filed in the United States before the invention by the applicant for patent or (2) a patent granted on an application for patent by another filed in the United States before the invention by the applicant for patent, except that an international application filed under the treaty defined in section 351(a) shall have the effects for purposes of this subsection of an application filed in the United States only if the international application designated the United States and was published under Article 21(2) of such treaty in the English language.

The following is a quotation of 35 U.S.C. § 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-22 are rejected under 35 U.S.C. 102(e) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Nero et al. No distinction is seen between the

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process disclosed by Nero et al., and that recited in applicant's claims. Nero et al. disclose a process for the recovery of unreacted ammonia from the effluent obtained from a reaction zone wherein oxygen, ammonia and a hydrocarbon are reacted at an elevated temperature in the presence of an amoxidation catalyst to produce an unsaturated nitrile, comprising the steps of quenching the effluent with an aqueous ammonia phosphate solution, and heating the solution to an elevated temperature sufficient to reduce the ammonium ion concentration and to generate a vaporous stream containing ammonia. (See the Abstract and pages 3-5.) Nero et al. teach in paragraph [0046] on page 3 that the process includes a stripping step of treating the ammonia phosphate quench solution with a stripping gas to remove substantially all of the acrylonitrile and other useful coalproducts from a second solution prior to the heating step, and that a gas such as propane, nitrogen or carbon monoxide may be used as the stripping gas.

Claims 1-22 are rejected under 35 U.S.C. 102(b) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Carlson '148. No distinction is seen between the process disclosed by Carlson '148, and that recited in applicant's claims. Carlson '148 discloses a process for recovering ammonia from a mixture of gases containing ammonia and hydrogen cyanide which comprises passing the mixture through an

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aqueous absorbing solution of ammonium hydrogen phosphate, whereby substantially all of the ammonia and a minor portion of the hydrogen cyanide are absorbed in the solution and a major portion of the hydrogen cyanide passes unabsorbed from the solution, subsequently passing through the solution resulting from the above step such an amount of steam as to remove absorbed hydrogen cyanide but only a minor proportion of absorbed ammonia therefrom, subjecting the resulting solution to heating to strip remaining available ammonia therefrom, and collecting the ammonia stripped from the solution in the last step. (See column 4, lines 23-36.) The steam employed as a stripping gas in the process of Carlson '148 would constitute a "stripping gas substantially free of carbon dioxide" as recited in applicant's claims.

Brazdil et al. '635, Scheibel, Rice et al. and Carlson et al. '731 are made of record for disclosing methods for recovering ammonia from gaseous mixtures containing ammonia with other gases.

Brazdil et al. '214 is made of record for disclosing the reaction of propane and isobutane with oxygen and ammonia to make unsaturated nitriles and olefins.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Wayne A. Langel whose telephone number is (703) 308-0248. The examiner

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can normally be reached on Monday through Friday from 8 A.M. to 3:30 P.M.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Stanley Silverman, can be reached on (703) 308-3837. The fax phone number for this Group is (703) 305-7718.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the Group receptionist whose telephone number is (703) 308-2351.

WAL:cdc

May 1, 2003

Mayne A. Jangel Wayne A. Langel PRIMARY EXAMINER